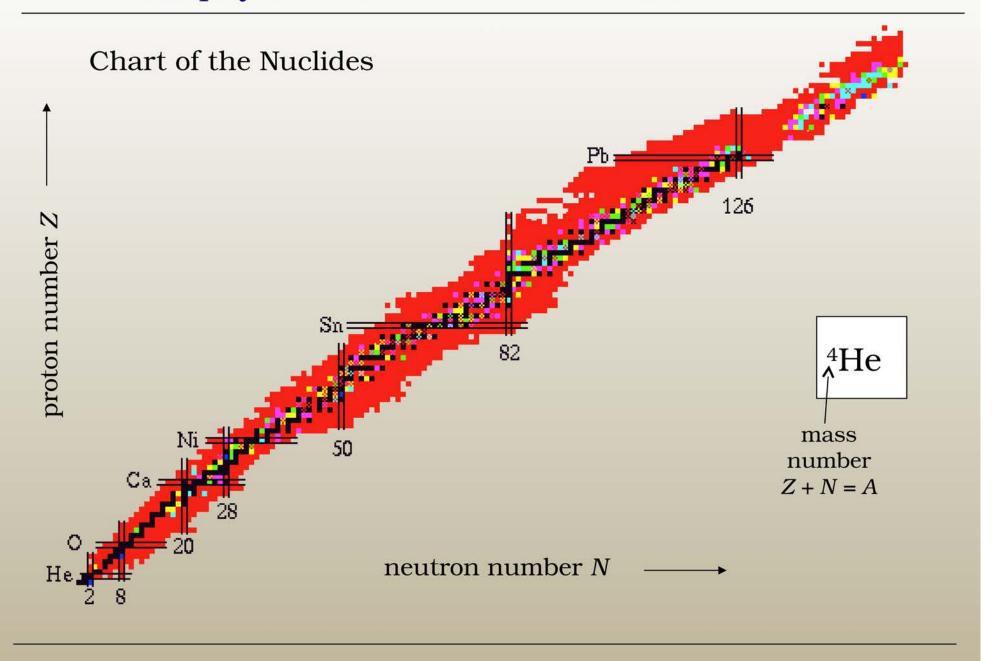
Nuclear Astrophysis Lecture 2: Nucleosynthesis

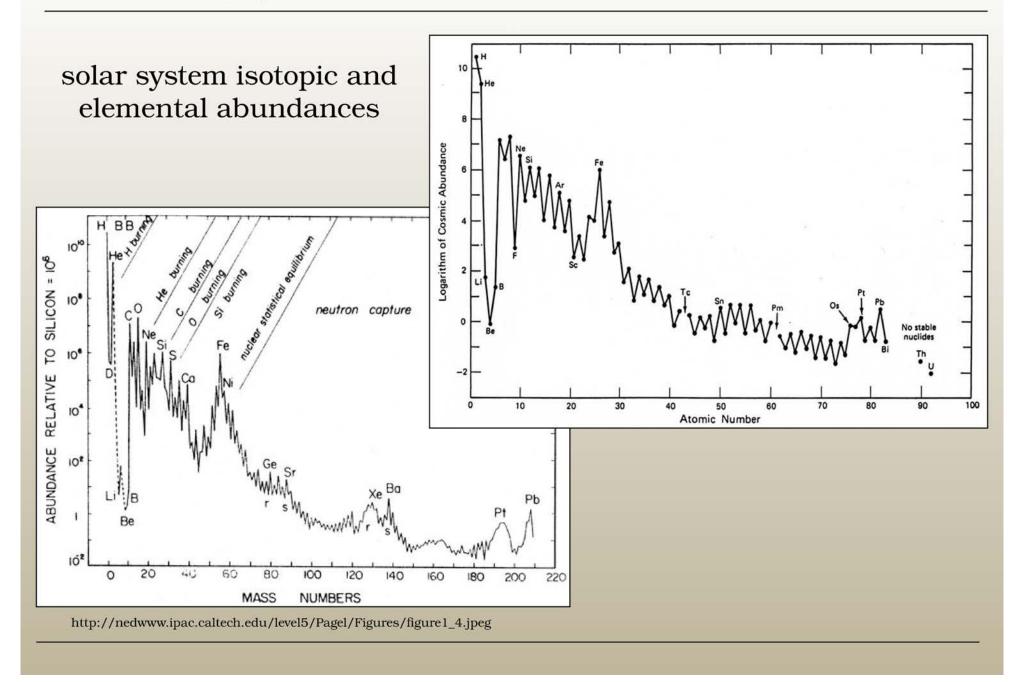
Gail McLaughlin

North Carolina State University

Objectives of Lecture 2

- Understand types of nucleosynthesis
- Understand where various types are made
- Be able to write down equations for a reaction network





some terminology

 n_i number of species j per unit volume

 W_i atomic weight (or molar mass) of species j

 ρ_m mass density

$$\rho_m = \frac{\sum_{j} n_j W_j}{N_A}$$

 X_i nucleon fraction, or mass fraction

$$X_{j} = \frac{n_{j} A_{j}}{\rho N_{A}}$$

Note
$$\sum_{j} X_{j} = \sum_{j} \frac{n_{j} A_{j}}{\rho N_{A}} = \frac{1}{\rho} \frac{\sum_{j} n_{j} A_{j}}{N_{A}} = 1$$

 ρ baryon mass density

$$\rho = \frac{\sum_{j} n_{j} A_{j}}{N_{A}}$$

 Y_i mole fraction, or abundance

$$Y_{j} = \frac{X_{j}}{A_{j}} = \frac{n_{j}}{\rho N_{A}}$$

some terminology

y_i rescaled abundances

In meteoritics, abundances are normally scaled relative to silicon

(set number of silicon atoms to be 10^6):

$$\log y_i = \log f_{Si} + \log Y_i$$

where f_{Si} is the appropriate normalizing constant. For $Y_{Si} = 2.529 \times 10^{-5}$, $\log f_{Si} = 10.5970$.

Astronomers sometimes use a scale relative to hydrogen

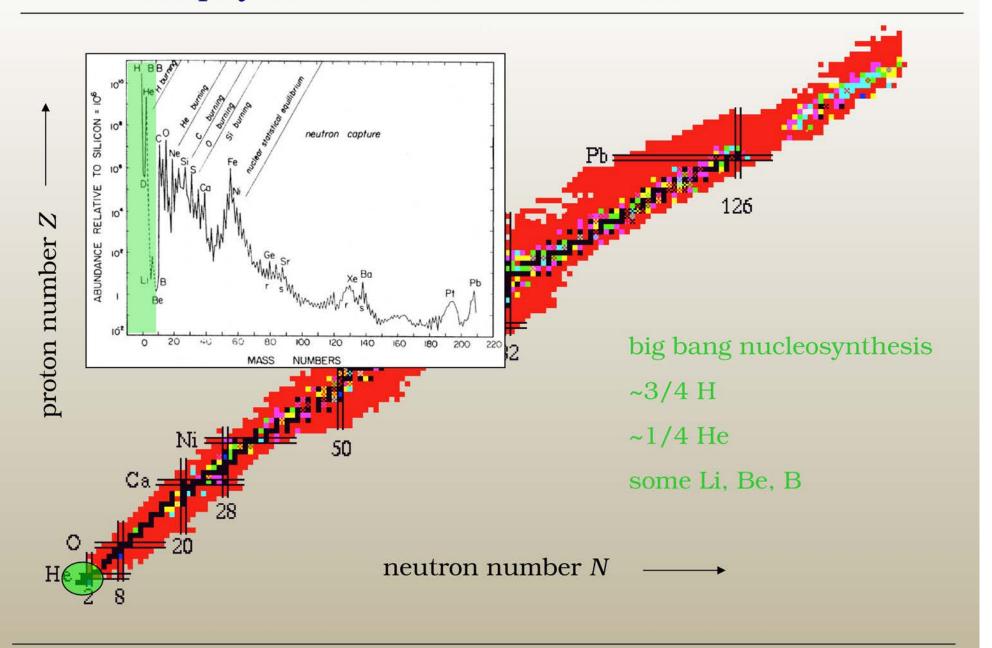
(set number of hydrogen atoms to be 10^{12}):

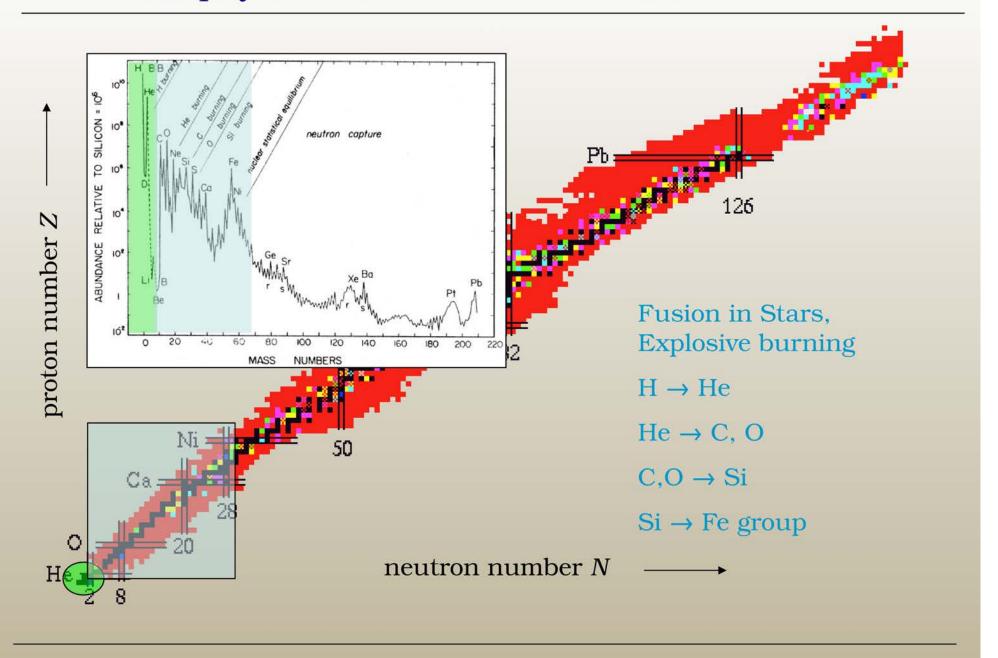
$$\log y_j = \log f_H + \log Y_i$$

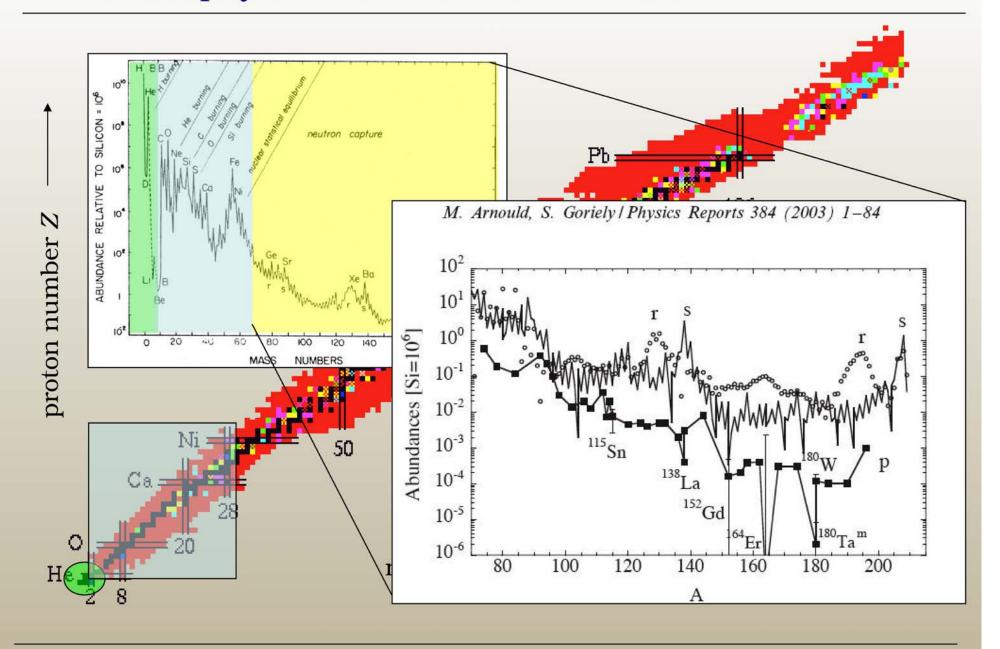
More commonly, astronomers express abundances as ratios relative to solar:

$$\log[n_i/n_j]_{\text{star}} - \log[n_i/n_j]_{\text{solar}} \equiv [i/j]$$

For example, a ratio of sodium to iron which is half solar would be [Na/Fe] = -0.3.







What information is needed?

My list:

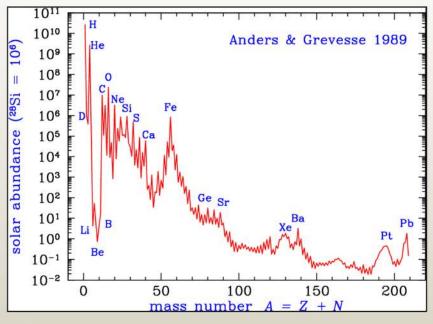
- astrophysical conditions, i.e. temperature and density
- nuclear physics input (masses, reaction rates, decay rates)

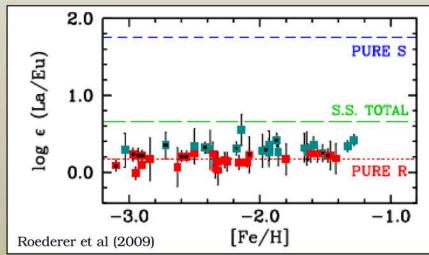
What data can you compare with?

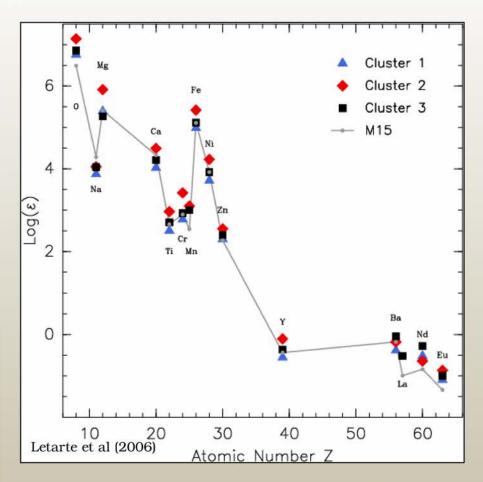
My list:

- solar system abundances (sun, earth)
- meterorites
- spectroscopic data from other stars

example abundance patterns







$$\log(\varepsilon_i) = \log_{10}(n_i/n_{\rm H}) + 12$$

some terminology

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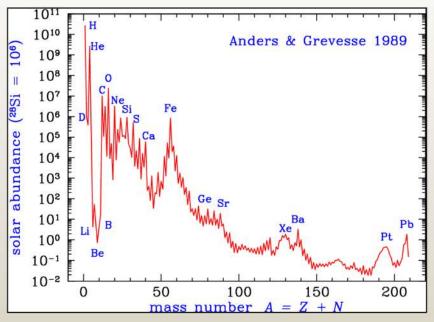
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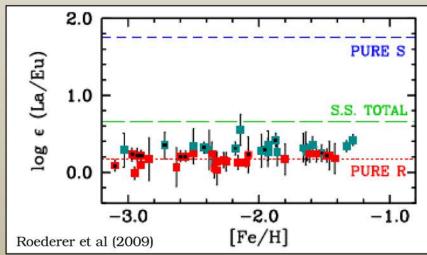
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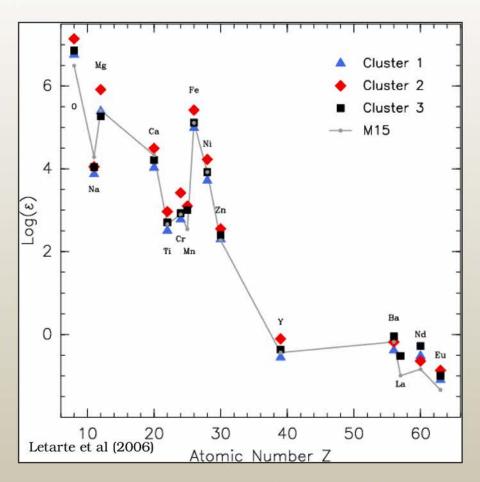
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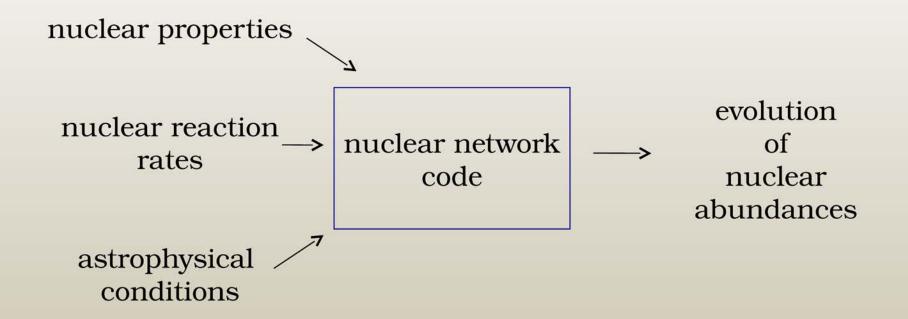
example abundance patterns





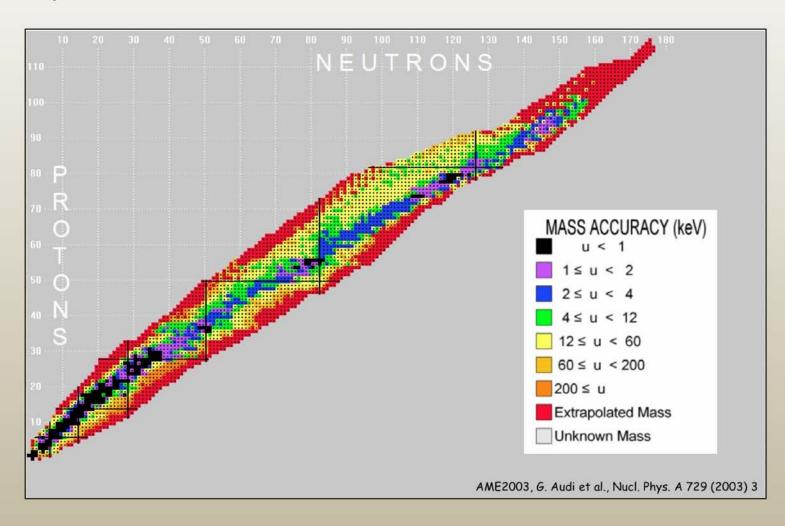


$$\log(\varepsilon_i) = \log_{10}(n_i/n_{\rm H}) + 12$$



building a nuclear network code: nuclear properties

atomic/nuclear masses



building a nuclear network code: nuclear properties

radioactive decay rates $N(t) = N_0 e^{-\lambda t}$

$$N(t) = N_0 e^{-\lambda t}$$

decay constant

mean lifetime, $\tau = 1/\lambda$

 $T_{1/2}$ half - life, $T_{1/2} = \ln 2/\lambda$

How determined?

Fermi's 'Golden Rule'

$$rate = \frac{2\pi}{h} \left| \left\langle f \middle| H_{int} \middle| i \right\rangle \right|^2 \rho(E)$$

final and initial state wavefunctions

weak interaction Hamiltonian $H_{\rm int}$

density of states for the final particles

building a nuclear network code: reaction rates

cross section for the reaction $i + j \rightarrow k + l$

$$\sigma_{ij}(v) = \frac{\text{number of reactions per nucleus } i \text{ per second}}{\text{flux of incoming projectiles } j}$$

$$\sigma_{ij}(v) = \frac{r_{ij}/n_i}{n_j v_{ij}}$$

 r_{ii} number of interactions i(j,k)l per second

 v_{ii} relative velocity of particles i, j

So the reaction rate per unit volume is just:

$$r_{ij} = n_i n_j v_{ij} \sigma_{ij}(v)$$

building a nuclear network code: reaction rates

In astrophysical environments the relative velocity v_{ij} is not constant, but instead there exists a distribution of relative velocities, which can be described by the probability function P(v), where:

$$\int_{0}^{\infty} P(v)dv = 1$$

So the reaction rate can be generalized to:

$$r_{ij} = n_i n_j \int_0^\infty v P(v) \sigma_{ij}(v) dv$$
$$r_{ij} = n_i n_j \langle \sigma v \rangle_{ij}$$

building a nuclear network code: reaction rates

If the nuclei are nonrelativistic and nondegenerate, their velocities can be described by a Maxwell-Boltzmann distribution

$$P(v)dv = \left(\frac{m_{ij}}{2\pi kT}\right)^{3/2} e^{-m_{ij}v^2/2kT} 4\pi v^2 dv$$

where:

 m_{ij} reduced mass, $m_{ij} = m_i m_j / (m_i + m_j)$

T temperature

k Boltzmann constant, $k = 8.6173 \times 10^{-5} \text{ eV/K}$

The velocity distribution can be written as an energy distribution, since $E = m_{ii}v^2/2$

$$P(v)dv = P(E)dE = \frac{2}{\sqrt{\pi}} \frac{1}{(kT)^{3/2}} \sqrt{E} e^{-E/kT} dE$$

building a nuclear network code

Now consider the rate of change in the number density of species *j*:

$$\frac{dn_{j}}{dt} = n_{k} n_{l} \langle \sigma v \rangle_{kl,j} - n_{j} n_{l} \langle \sigma v \rangle_{jl,n} + n_{i} \lambda_{i,j} - n_{j} \lambda_{j,m} + K$$

Note for reactions involving identical particles, a term of the form:

$$\frac{n_i^2}{2!} \langle \sigma v \rangle_{ii,j}$$
 (two body) or $\frac{n_i^3}{3!} \langle \sigma v \rangle_{iii,j}$ (three body)

is needed.

The above can be written in terms of abundances as:

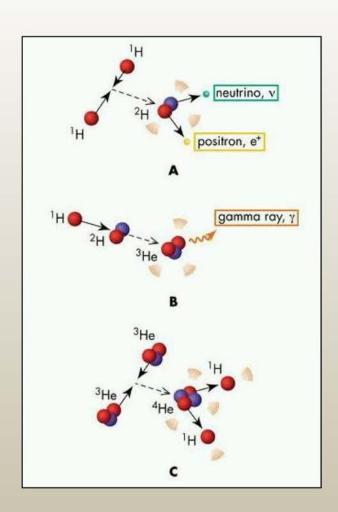
$$\frac{dY_{j}}{dt} = Y_{k}Y \rho N_{A} \langle \sigma v \rangle_{kl,j} - Y_{j}Y_{l}\rho N_{A} \langle \sigma v \rangle_{jl,n} + Y_{i}\lambda_{i,j} - Y_{j}\lambda_{j,m} + K$$

this is often what we call the reaction rate in a network code

the basic proton-proton chain (PPI)

Conversion of ¹H to ⁴He via:

$${}^{1}\text{H} + {}^{1}\text{H} \rightarrow {}^{2}\text{H} + e^{+} + v_{e}$$
 ${}^{2}\text{H} + {}^{1}\text{H} \rightarrow {}^{3}\text{He} + \gamma$
 ${}^{3}\text{He} + {}^{3}\text{He} \rightarrow {}^{4}\text{He} + {}^{1}\text{H} + {}^{1}\text{H}$



PPI network equations

Recall the abundances evolve as

$$\frac{dY_{j}}{dt} = Y_{k}Y_{l}\rho N_{A}\langle \sigma v \rangle_{kl,j} - Y_{j}Y_{l}\rho N_{A}\langle \sigma v \rangle_{jl,n} + Y_{i}\lambda_{i,j} - Y_{j}\lambda_{j,m} + K$$

So here we have a system of four differential equations:

$$\begin{split} \frac{dY_{\rm H}}{dt} &= 2{Y_{^{3}}}_{\rm He}^{\ 2}\rho N_{A}\langle\sigma v\rangle_{^{3}}_{\rm He^{^{3}}He,^{^{4}}He} - {Y_{\rm H}}^{2}\rho N_{A}\langle\sigma v\rangle_{\rm HH,D} - {Y_{\rm D}}Y_{\rm H}\rho N_{A}\langle\sigma v\rangle_{\rm DH,^{^{3}}He} \\ \frac{dY_{\rm D}}{dt} &= \frac{1}{2}{Y_{\rm H}}^{2}\rho N_{A}\langle\sigma v\rangle_{\rm HH,D} - {Y_{\rm D}}Y_{\rm H}\rho N_{A}\langle\sigma v\rangle_{\rm DH,^{^{3}}He} \\ \frac{dY_{^{3}}_{\rm He}}{dt} &= {Y_{\rm D}}Y_{\rm H}\rho N_{A}\langle\sigma v\rangle_{\rm DH,^{^{3}}He} - {Y_{^{3}}}_{\rm He}^{\ 2}\rho N_{A}\langle\sigma v\rangle_{^{3}}_{\rm He^{^{3}}He,^{^{4}}He} \\ \frac{dY_{^{4}}_{\rm He}}{dt} &= \frac{1}{2}{Y_{^{3}}}_{\rm He}^{\ 2}\rho N_{A}\langle\sigma v\rangle_{^{3}}_{\rm He^{^{3}}He,^{^{4}}He} \end{split}$$

PPI network equations

But note first D, then ³He will come into steady-state:

$$\frac{dY_{\rm D}}{dt} = \frac{1}{2} Y_{\rm H}^2 \rho N_A \langle \sigma v \rangle_{\rm HH,D} - Y_{\rm D} Y_{\rm H} \rho N_A \langle \sigma v \rangle_{\rm DH,^3 He} = 0, \text{ so}$$

$$\frac{Y_{\rm D}}{Y_{\rm H}} = \frac{\langle \sigma v \rangle_{\rm HH,D}}{2 \langle \sigma v \rangle_{\rm DH,^3 He}} \sim 10^{-17}$$

$$\frac{dY_{^{3}\text{He}}}{dt} = Y_{D}Y_{H}\rho N_{A}\langle\sigma\nu\rangle_{DH,^{3}\text{He}} - Y_{^{3}\text{He}}^{2}\rho N_{A}\langle\sigma\nu\rangle_{^{3}\text{He}^{3}\text{He},^{4}\text{He}}
\frac{dY_{^{3}\text{He}}}{dt} = \frac{1}{2}Y_{H}^{2}\rho N_{A}\langle\sigma\nu\rangle_{HH,D} - Y_{^{3}\text{He}}^{2}\rho N_{A}\langle\sigma\nu\rangle_{^{3}\text{He}^{3}\text{He},^{4}\text{He}} = 0
\frac{Y_{^{3}\text{He}}}{Y_{H}} = \sqrt{\frac{\langle\sigma\nu\rangle_{HH,D}}{2\langle\sigma\nu\rangle_{^{3}\text{He}^{3}\text{He},^{4}\text{He}}} \sim 10^{-5}$$

Higher T than the sun; nuclear statistical equilibrium

When strong and electromagnetic interactions come into equilibrium at high temperatures, the nuclear abundances are no longer sensitive to individual reaction rates and only depend on the temperature T, density ρ , and the neutron-richness of the composition.

Nuclear statistical equilibrium (NSE) abundances are given by:

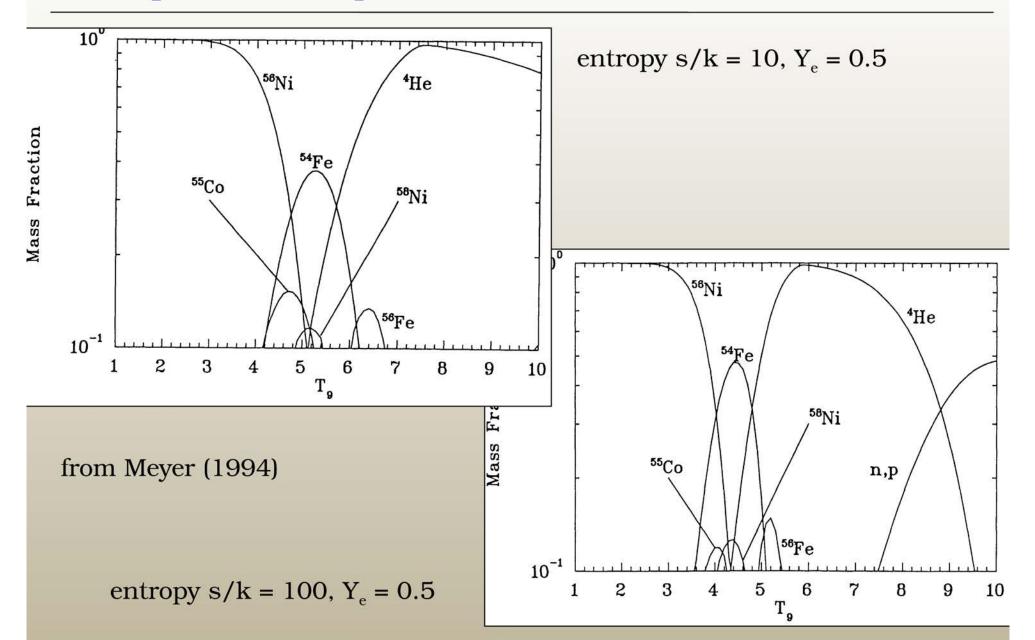
$$Y_{i} = (\rho N_{A})^{A_{i}-1} \frac{G_{i}}{2^{A_{i}}} A_{i}^{3/2} \left(\frac{2\pi h^{2}}{m_{u}kT}\right)^{\frac{3}{2}(A_{i}-1)} \exp\left[\frac{B_{i}}{kT}\right] Y_{p}^{Z_{i}} Y_{n}^{N_{i}}$$

where G_i is the nuclear partition function and B_i is the binding energy.

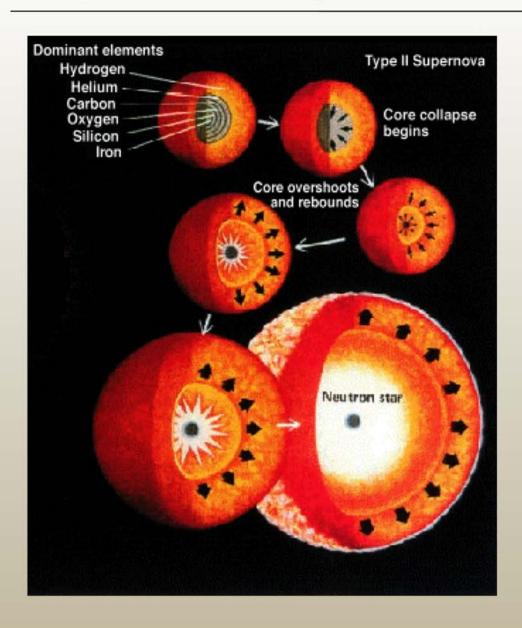
We also require mass and charge conservation:

$$\sum_{i} Y_{i} A_{i} = 1 \qquad \sum_{i} Y_{i} Z_{i} = Y_{e}$$
electron fraction

sample NSE compositions

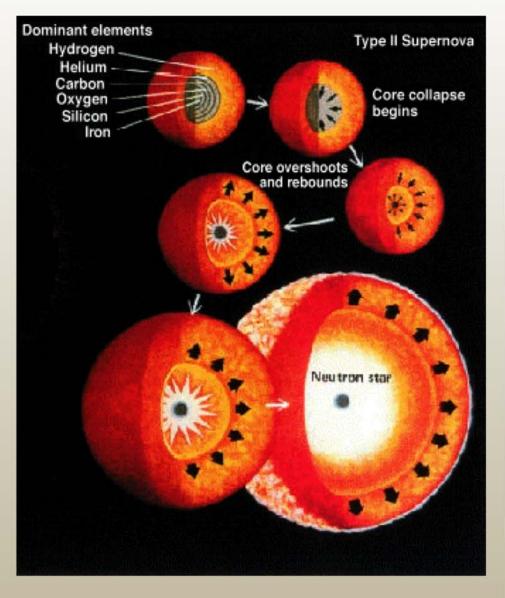


explosive burning in core-collapse supernovae

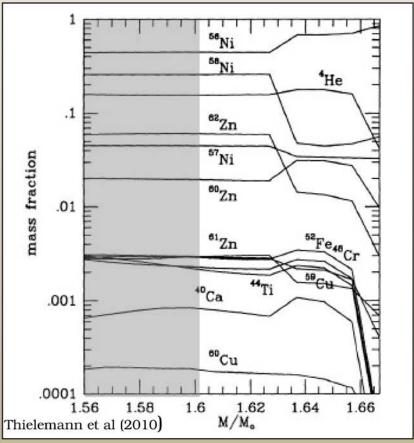


Outgoing shock wave heats inner Si and O layers – NSE is achieved and rapidly freezes out as shock passes

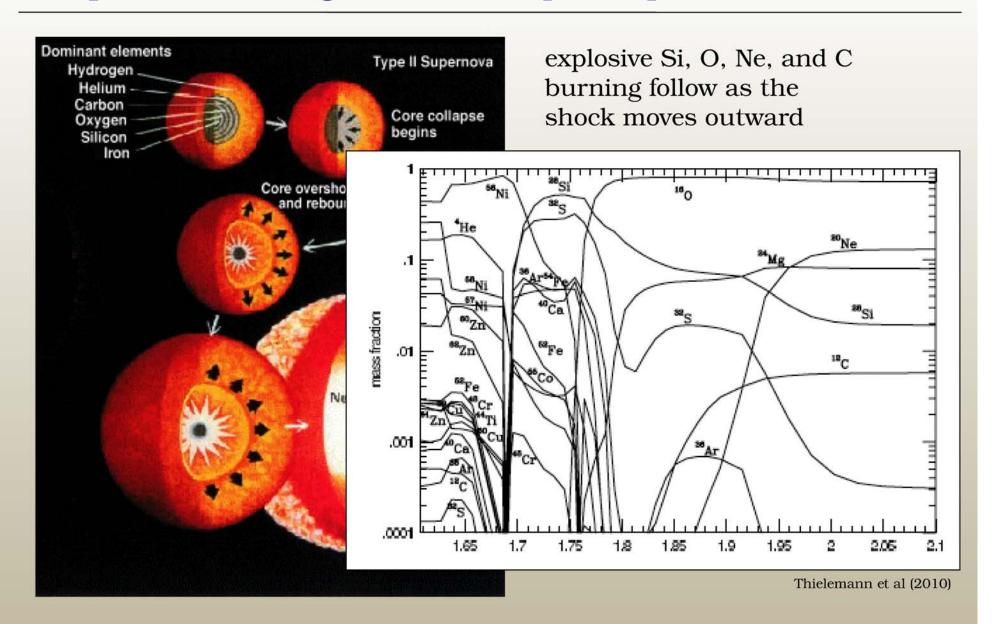
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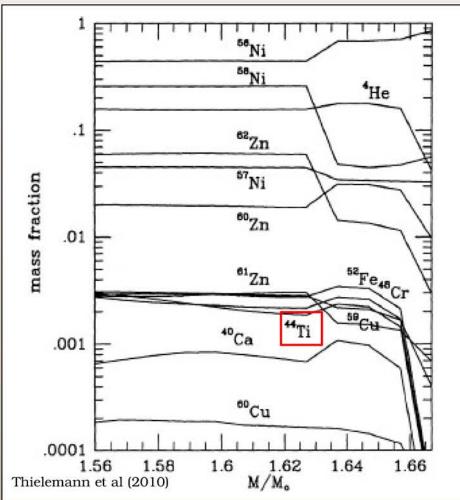


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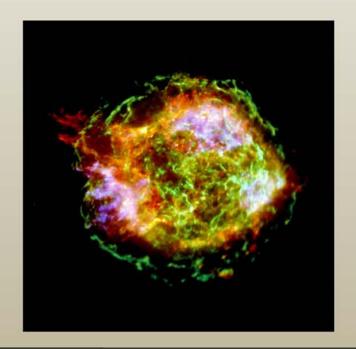


SNe nucleosynthesis – long-lived radionuclides

44Ti created in alpha-rich freezeout from NSE, close to the mass cut



 $T_{1/2}$ ~ 60 years x-rays from the decay chain observed in SNe remnants Cas-A and RX J0852.0-4622

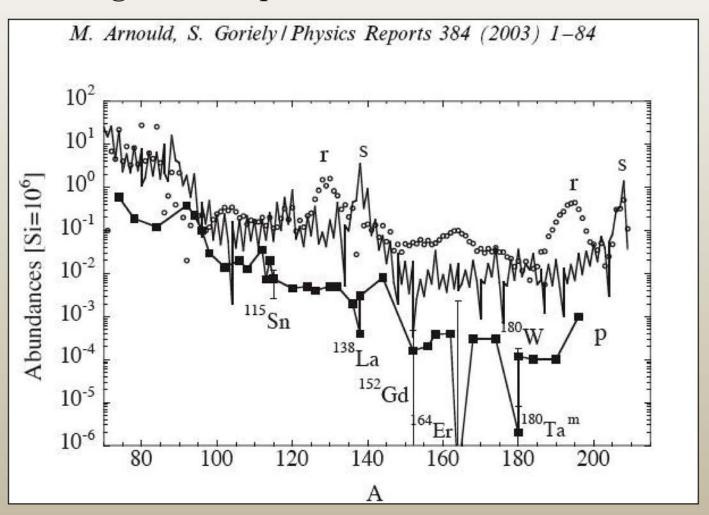


What about the heavier elements?

- p-process
- s-process
- r-process

SNe nucleosynthesis – proton-rich heavy elements

heavy p-process nuclei are made by (γ,n) photodissociations of pre-existing r- and s- process nuclei

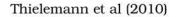


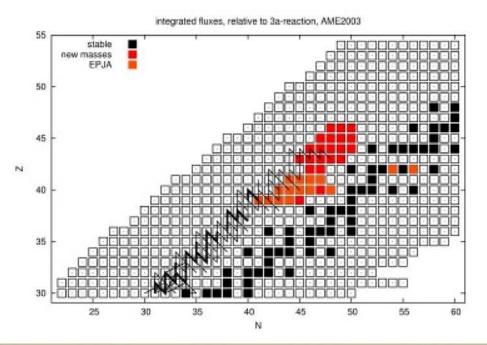
SNe nucleosynthesis – proton-rich heavy elements

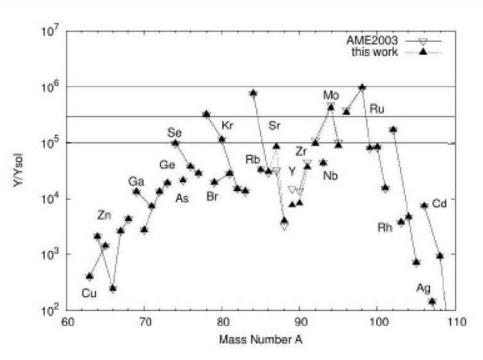
vp-process

(Frohlich et al 2006, Pruet et al 2006, Wanajo 2006)

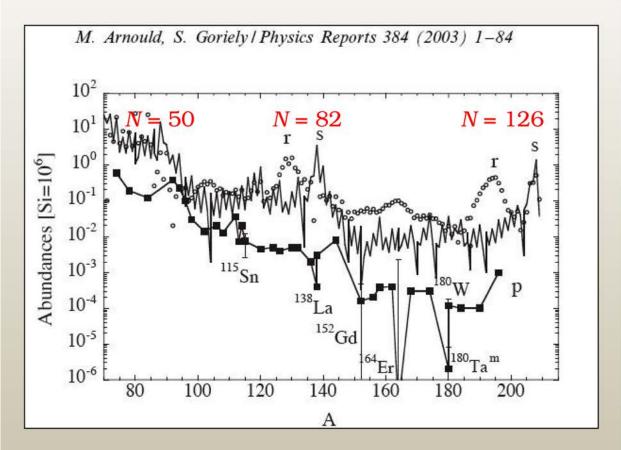
Thought to occur in proton-rich ejecta from the inner regions of the SNe $p + \overline{V}_e \rightarrow n + e^+$

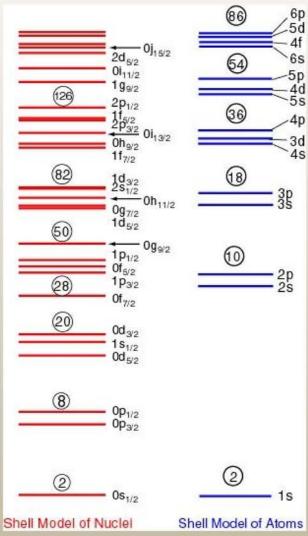




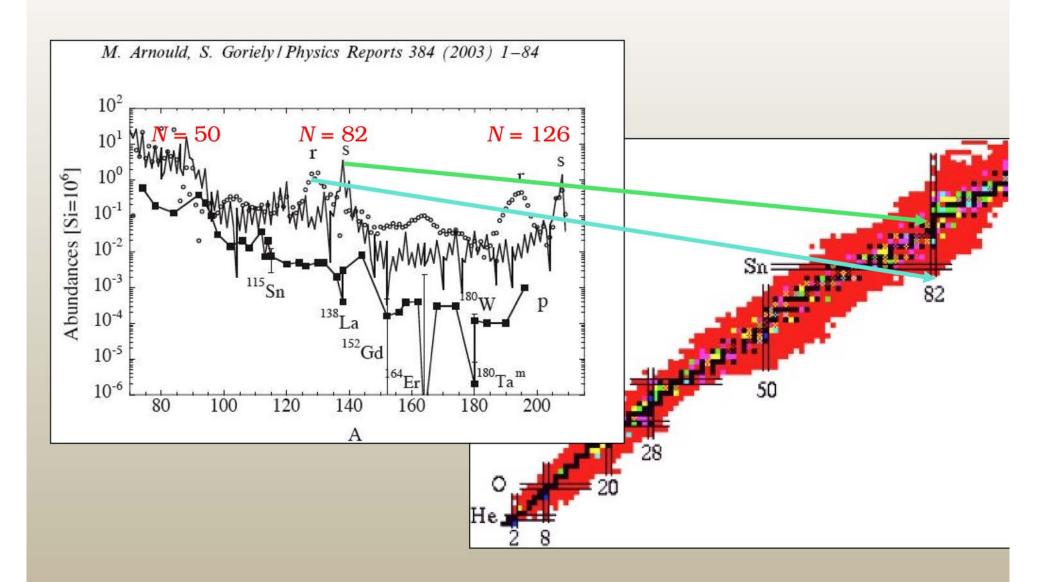


neutron capture nucleosynthesis

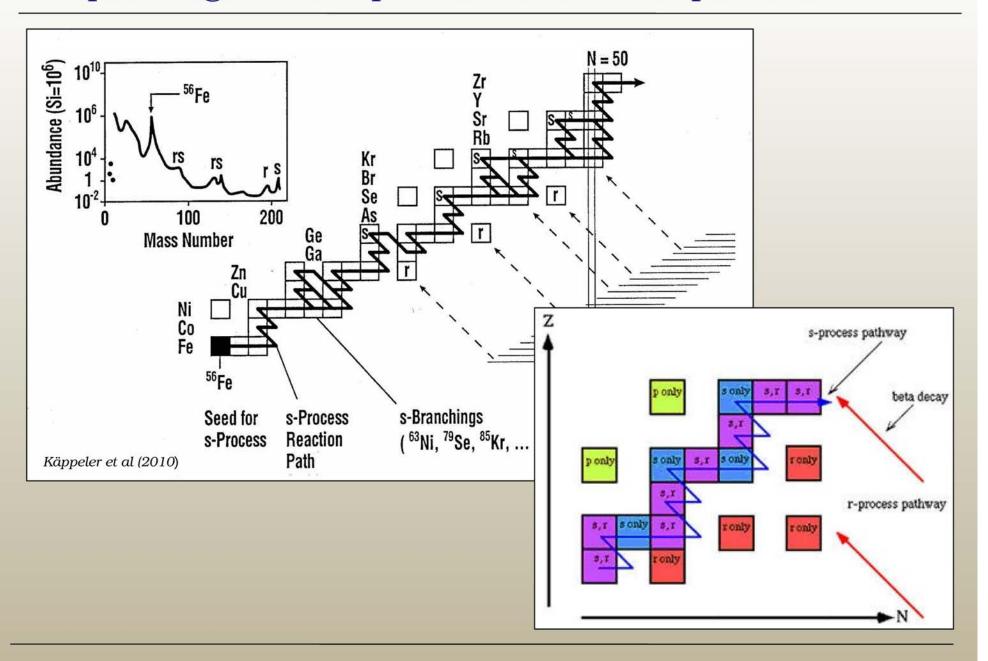




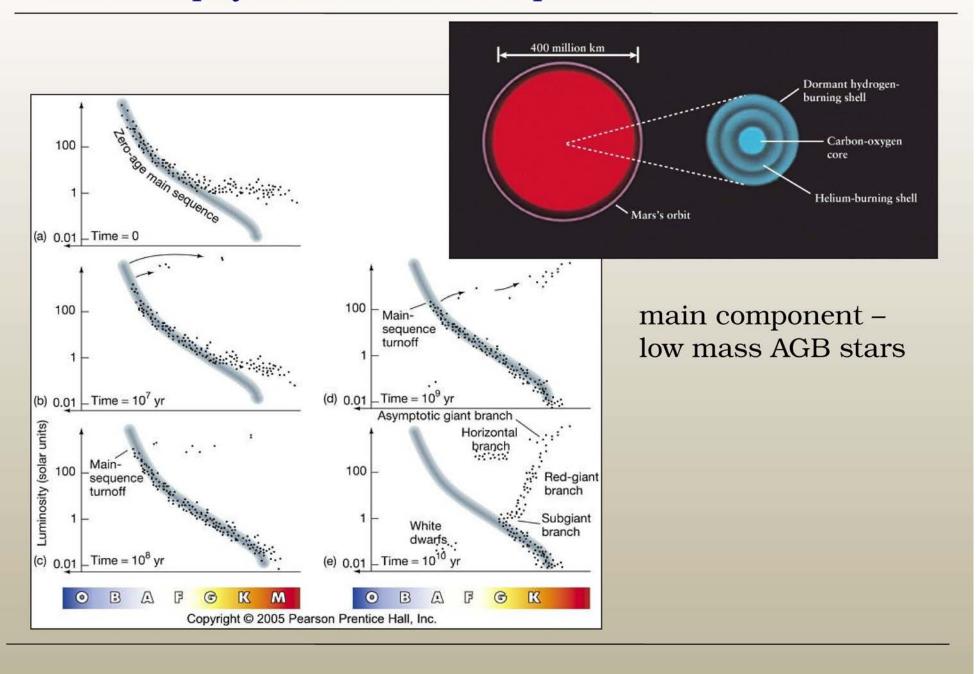
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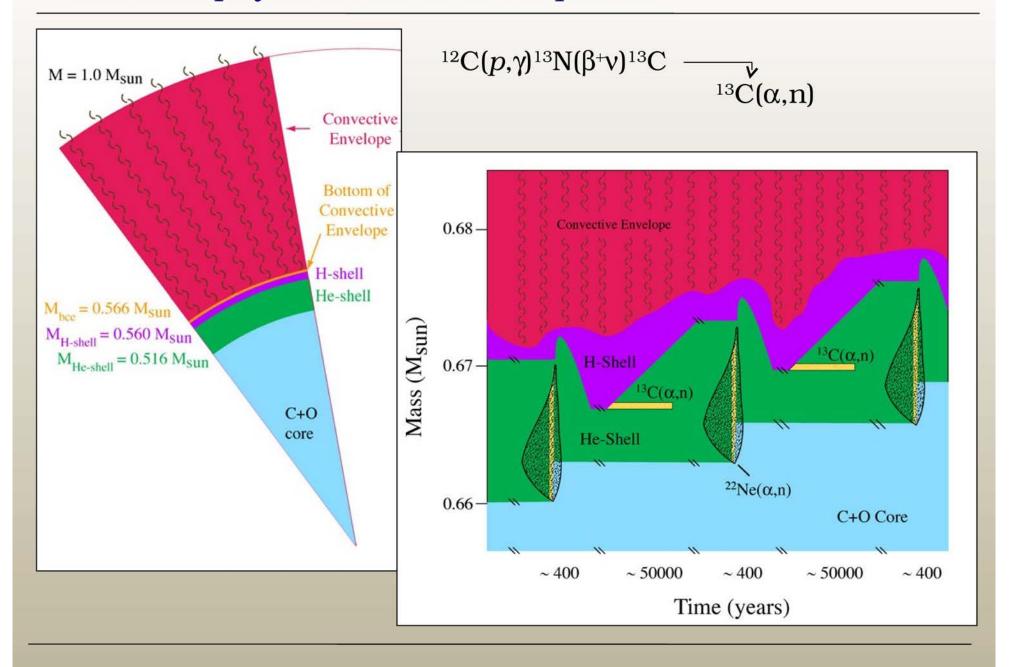
separating *s*- and *r*-process abundance patterns



the astrophysical site of the s-process



the astrophysical site of the s-process



Next lecture: r-process